

perature for 3.5 h. The reaction mixture was neutralized and cooled overnight at 5 °C to allow precipitation of product. The product was collected by vacuum filtration and then lyophilized to yield 0.330 g (1.1 mmol, 41%) of ethenoguanosine (18). The spectroscopic data of 18 were consistent with the literature values.³⁰

Reaction of Guanosine (14) with Glycidaldehyde (1). Guanosine (1.942 g, 6.9 mmol) was placed in H₂O (200 mL), and the solution was basified to pH 10 with warming to aid dissolution. Glycidaldehyde (1) 0.57 g (7.9 mmol) was added, and the reaction mixture was stirred at room temperature for 1 h. The reaction mixture was neutralized and cooled to allow precipitation of white crystals. This material was suspended in water and lyophilized to give 1.364 g (4.0 mmol) of 5,9-dihydro-7-(hydroxymethyl)-9-oxo-3-β-D-ribofuranosyl-3*H*-imidazo[1,2-*a*]purine (17) as white crystals in 59% yield: mp >300 °C dec; UV (0.1 N HCl) λ_{max} 300 nm (ε 7.7 × 10³), 276 (9.9 × 10³), 226 (2.54 × 10⁴), (pH 7 buffer) 285 (9.9 × 10³), 228 (2.63 × 10⁴), (0.1N NaOH) 310 (7.4 × 10³), 285 (6.6 × 10³), 238 (2.85 × 10⁴); mass spectrum, *m/z* (relative intensity) 207 (2.6), 189 (75.9, "base" + 2 H - OH), 188 (36.5), 133 (7.1); ¹H NMR (Me₂SO-*d*₆) δ 3.53-3.65 (m, 2 H), 3.91 (m, 1 H), 4.12 (m, 1 H), 4.45 (m, 1 H), 4.84 (d, 2 H, *J* = 6.0 Hz), 4.98 (t, 1 H, *J* = 6.0 Hz), 5.05 (t, 1 H, *J* = 5.3 Hz), 5.16 (d, 1 H, *J* = 4.7 Hz), 5.42 (d, 1 H, *J* = 6.0 Hz), 5.81 (d, 1 H, *J* = 5.9 Hz), 7.23 (s, 1 H), 8.14 (s, 1 H), 12.36 (s, 1 H); ¹³C NMR (Me₂SO-*d*₆) δ 55.0, 61.2, 70.2, 73.6, 85.1, 86.8, 113.7, 115.9, 124.6, 137.3, 146.7, 150.2, 153.6.

Anal. Calcd for C₁₃H₁₅N₅O₆·H₂O: C, 43.95; H, 4.82; N, 19.71. Found: C, 44.37; H, 4.48; N, 20.20.

Reaction of Adenosine (19) with Glycidaldehyde (1) at pH 5. Adenosine (19) (0.280 g, 1.0 mmol) was allowed to react with glycidaldehyde (1) (0.122 g, 1.7 mmol) for 36 h at pH 5. Separation yielded 0.090 g (0.3 mmol, 30%) of 3-β-D-ribofuranosyl-7-(hydroxymethyl)-3*H*-imidazo[2,1-*i*]purine (20) as bluish transparent crystals: mp 214-216 °C; UV (H₂O) λ_{max} 231 nm (ε 2.9 × 10⁴), 268 (6.7 × 10³), 279 (6.4 × 10³), 300 (3.0 × 10³); mass spectrum, *m/z* (relative intensity) 321 (M⁺ + 4.2), 189 ("base" + H, 100.0), 188 ("base", 22.9), 172 ("base" + H - OH, 91.8), 135 (50.7), 133 (14.7); ¹H NMR (Me₂SO-*d*₆) δ 3.67 (m, 2 H), 4.00 (m, 1 H), 4.22 (t, 1 H, *J* = 4.4 Hz), 4.61 (t, 1 H, *J* = 4.9 Hz), 4.91 (s, 2 H), 5.1-5.5 (brs, 4 H), 6.08 (d, 1 H, *J* = 5.4 Hz), 7.48 (s, 1 H), 8.59 (s, 1 H), 9.15 (s, 1 H).

Anal. Calcd for C₁₃H₁₅N₅O₅: C, 48.60; H, 4.71; N, 21.80. Found: C, 48.70; H, 4.83; N, 21.69.

Reaction of Adenosine (19) with 2,3-Epoxybutanal (2) at pH 5. Adenosine (19) (0.282 g, 1.1 mmol) was stirred for 48 h with 0.094 g (1.1 mmol) of 2,3-epoxybutanal (2). Separation

yielded 0.022 g (0.07 mmol, 6% yield, 7% conversion based on unreacted adenosine) of 21 as fluffy white crystals: mp 219-221 °C; UV (H₂O) λ_{max} 231 nm (ε 3.00 × 10⁴), 268 (5.9 × 10³), 279 (5.9 × 10³), 300 (sh, 3.1 × 10³); mass spectrum, *m/z* (relative intensity) 203 ("base" + H, 14.2), 188 ("base" + H - CH₃, 25.0), 159 ("base" - C₂H₄O, 100.0); ¹H NMR (Me₂SO-*d*₆) δ 1.64 (d, 3 H, *J* = 6.4 Hz), 3.68 (m, 2 H), 4.00 (d, 1 H, *J* = 3.4 Hz), 4.19 (m, 1 H), 4.60 (m, 1 H), 5.24-5.05 (m, 3 H), 5.51 (m, 2 H), 6.07 (d, 1 H, *J* = 5.9 Hz), 7.44 (s, 1 H), 8.57 (s, 1 H), 9.19 (s, 1 H).

Anal. Calcd for C₁₄H₁₇N₅O₅·H₂O: C, 47.59; H, 5.42; N, 19.82. Found: C, 48.02; H, 5.43; N, 19.25.

Preparation of 9-Ethyl-1, *N*⁶-ethenoadenine-10-carboxaldehyde (23). This compound was prepared as described previously²² and was obtained in 38% yield as white crystals: mp 223-225 °C; UV (95% ethanol) λ_{max} 230 nm (ε 2.06 × 10⁴), 328 (1.51 × 10⁴), 339 (1.50 × 10⁴); mass spectrum, *m/z* (relative intensity) 216 (M⁺ + 1, 12.2), 215 (M⁺, 100.0), 187 (M⁺ - CO, 34.4), 186 (M⁺ - C₂H₅, 34.4); ¹H NMR (CDCl₃) δ 1.62 (t, 3 H), 4.44 (q, 2 H), 8.13 (s, 1 H), 8.37 (s, 1 H), 10.02 (s, 1 H), 10.08 (s, 1 H).

Reduction of 9-Ethyl-1, *N*⁶-ethenoadenine-10-carboxaldehyde (23). To a solution of 0.042 g (1.1 mmol) of NaBH₄ in 20 mL of cold ethanol was added 0.052 g (0.24 mmol) of 9-ethyl-1, *N*⁶-ethenoadenine-10-carboxaldehyde (23). The reaction was stirred for 1/2 h at room temperature and then the solvent removed in vacuo. Separation on silica gel preparative layer plates with 13% MeOH/CHCl₃ yielded 0.023 g (0.11 mmol, 46%) of 3-ethyl-7-(hydroxymethyl)-3*H*-imidazo[2,1-*i*]purine (22) as off-white crystals: mp 195 °C dec; UV (H₂O) λ_{max} 233 nm (ε 2.7 × 10⁴), 269 (6.5 × 10³), 279 (9.2 × 10³), 300 (3.9 × 10³); mass spectrum *m/z* (relative intensity) 218 (M⁺ + 1, 6.7), 217 (M⁺, 50.4), 200 (M⁺ - OH, 100.0), 172 (39.4); ¹H NMR (Me₂SO-*d*₆) δ 1.48 (t, 3 H, *J* = 7.3 Hz), 4.35 (q, 2 H, *J* = 7.3 Hz), 4.91 (m, 2 H), 5.34 (m, 1 H), 7.44 (s, 1 H), 8.33 (s, 1 H), 9.11 (s, 1 H).

Anal. Calcd for C₁₀H₁₁N₅O·H₂O: C, 51.05; H, 5.57; N, 29.77. Found: C, 51.22; H, 5.70; N, 29.13.

Acknowledgment. Support of these investigations by a grant (CHE-8200818) from the NSF is gratefully acknowledged. The Bruker WM-360 high-field NMR spectrometer was purchased in part with a grant (CHE-8201836) from the NSF.

Registry No. 1, 765-34-4; 2, 3209-33-4; 3, 1122-47-0; 4, 65-46-3; 5, 99310-29-9; 6, 99310-30-2; 7, 99310-31-3; 8, 4401-11-0; 9, 99310-32-4; 10, 90754-37-3; 14, 118-00-3; 17, 89647-27-8; 18, 62462-38-8; 19, 58-61-7; 20, 99310-33-5; 21, 99310-34-6; 22, 99310-35-7; 23, 91898-80-5.

Exserohilone: A Novel Phytotoxin Produced by *Exserohilum holmii*

Koko Sugawara,¹ Fumio Sugawara,² and Gary A. Strobel*

Department of Plant Pathology, Montana State University, Bozeman, Montana 59717

Yali Fu, He Cun-Heng, and Jon Clardy*

Department of Chemistry—Baker Laboratory, Cornell University, Ithaca, New York 14853-1301

Received September 18, 1985

A novel phytotoxin, exserohilone (1), was isolated from the culture broth of *Exserohilum holmii*, a pathogenic fungus of the weedy plant *Dactyloctenium aegyptium*. The structure of exserohilone (1) was elucidated by X-ray diffraction analysis of the bis(*p*-bromobenzoate) derivative 3. 9,10-Dihydroexserohilone (2) was also isolated, and its structure was determined by spectral methods.

Bacterial and fungal pathogens of plants often produce disease symptoms by elaborating phytotoxins in the host.³ There are relatively few studies on phytotoxins affecting

weedy plants, but such compounds could be useful herbicides,⁴ or serve as models for new herbicides.³ *Exserohilum holmii* is a fungal pathogen on *Dactyloctenium aegyptium* (crowfoot grass) which is a serious grasseous

(1) Present address: Bristol-Meyers Research Institute-Tokyo, 2-9-3, Shimo-meguro, Meguro-ku, Tokyo 153, Japan.

(2) Present address: RIKEN The Institute of Physical and Chemical Research, Wako-shi, Saitama 351-01, Japan.

(3) Strobel, G. A. *Ann. Rev. Biochem.* 1982, 51, 309-329.

(4) (a) Meigi Seika-Kaisha, U.S. Patent 4 448 601. (b) Tachibana, K.; Watanabe, T.; Sekisawa, Y.; Konnai, H.; Takematsu, T. *Pestic. Chem. Hum. Welfare Environ. Proc. Int. Congr. Pestic. Chem.*, 5th 1982, IV-19.

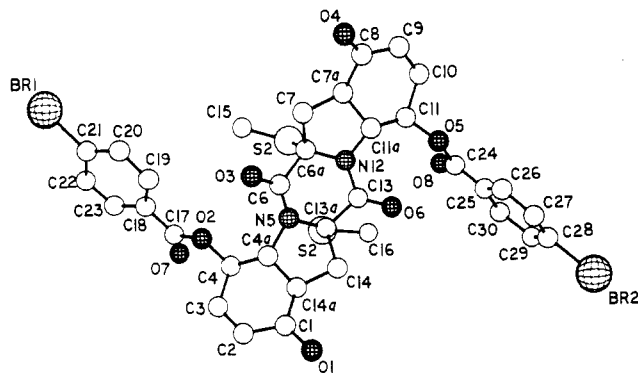
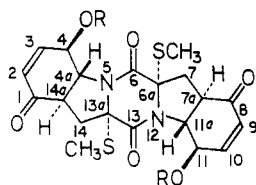


Figure 1. Computer-generated perspective drawing of the 4,11-bis(*p*-bromobenzoate) of exserochilone (3). Hydrogens are omitted for clarity, and no absolute configuration is implied.

weed in all major tropical and semitropical agricultural areas of the world. From cultures of this fungus we isolated two novel phytotoxins which are the subject of this report.

The well-split ^1H NMR of exserochilone (1) suggested that its structure consisted of rigid rings, which was also supported by its large optical rotation (-247°). The exchangeable signal (D_2O) at δ 6.10 (sharp s) disappeared by either acetylation or bromobenzoylation. This extremely low-field alcohol proton was attributed to strong hydrogen bonding with a ketone moiety. ^{13}C NMR revealed the presence of hydroxycyclohexenone [δ 195.2 (s), 128.6 (d), 151.5 (d), 72.9 (d)], methyl [δ 15.1 (q)], and methylene [δ 32.1 (t)] carbons. Decoupling experiments permitted us to draw the following partial structure: $\text{OCCH}=\text{CHCH}(\text{OH})\text{CH}(\text{X})\text{CH}(\text{Y})\text{CH}_2\text{Z}$. Of course, both the ^{13}C and ^1H NMR spectra were misleading in that they showed one-half the number of carbons and hydrogens present in the real structure. The structure was elucidated through a single-crystal X-ray diffraction analysis of the 4,11-bis(*p*-bromobenzoate) of 1 (3).



1, R = H
3, R = *p*-Br-C₆H₄CO

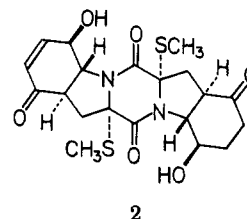
Compound 3 crystallized in the monoclinic space group $P2_1$ with $a = 11.289$ (2) Å, $b = 11.224$ (2) Å, $c = 14.904$ (2) Å, and $\beta = 63.91$ (1) $^\circ$ with one molecule of composition $\text{C}_{34}\text{H}_{28}\text{Br}_2\text{N}_2\text{O}_8\text{S}_2$ forming the asymmetric unit. All unique diffraction maxima with $2\theta \leq 114^\circ$ were collected on a computer controlled four-circle diffractometer with graphite monochromated $\text{Cu K}\alpha$ radiation (1.54178 Å) and variable speed 1° ω -scans. Of the 2430 reflections surveyed in this fashion, 2059 (85%) were judged observed ($F_o \geq 3\sigma(F_o)$) after correction for Lorentz, polarization, and background effects. A phasing model was found by locating the bromine positions in the Patterson synthesis and extending the structure with tangent formula recycling. Hydrogen atoms were located on a ΔF -synthesis or included at calculated positions. The final crystallographic discrepancy index was 0.047 for the observed reflections. Additional crystallographic details are available and are described in the paragraph entitled Supplementary Material at the end on this paper.

A computer-generated perspective drawing of the final X-ray model is given in Figure 1. Hydrogens are omitted for clarity, and the enantiomer shown is an arbitrary

choice. The structure is a twofold symmetric diketopiperazine and bond distances and angles agree well with generally accepted values.

Mass spectroscopy was troublesome even when FAB or CI was used; however finally, EIHRMS gave results consistent with the X-ray analysis as $\text{C}_{20}\text{H}_{22}\text{N}_2\text{O}_6\text{S}_2$ (M^+) at m/z 450.0919 and as $\text{C}_{19}\text{H}_{19}\text{N}_2\text{O}_6\text{S}_1$ ($\text{M}^+ - \text{SMe}$) at m/z 403.0963. The elucidated structure is closely related to that of spicorazine A, which is the antibiotic metabolite from *Epicoccum nigrum*.⁵

The EILRMS of 9,10-dihydroexserohilone (2) (colorless glass), gave a very similar fragmentation pattern to 1, but it was larger than 1 by 2 daltons. EIHRMS showed the molecular formula of 2 to be $\text{C}_{20}\text{H}_{24}\text{N}_2\text{O}_6\text{S}_2$ (M^+) at m/z 452.1075 and $\text{C}_{19}\text{H}_{21}\text{N}_2\text{O}_6\text{S}_1$ ($\text{M}^+ - \text{SMe}$) at m/z 405.1120. These MS data suggested that the structure of 2 was a saturated derivative of 1 i.e., a cyclohexenone had been reduced to a cyclohexanone. ^1H NMR supported this



2

formulation. The spectrum of 2 is more complex than that of 1, and it does not appear to be a symmetrical structure. For instance, the signals of the unsaturated side of the molecule are similar to those of 1, but the remaining signals are shifted upfield. Specifically the signals at δ 6.10 (OH), 4.70 (H-11), 3.89 (H-11a), 2.67 (H \bar{a} -7), and 2.23 (H \bar{a} -7) are shifted to δ 5.66 (OH), 4.27 (H-11), 3.77 (H-11a), 2.51 (H-7'), and 1.96 (H-7), respectively. Decoupling experiments were completely consistent with these assignments, and 2 is most plausibly the dihydro derivative of 1 shown.

Compounds 1 and 2 show nonselectivity toward several plant species at 10^{-4} – 10^{-5} M. These species include both crop and weedy plants when tests were performed by the leaf puncture bioassay method.⁶ These compounds cause the production of necrotic lesions usually surrounded by a reddish brown border. Compounds 1 and 2 may possess antimicrobial or antiviral activity since structurally related compounds such as aranotone,⁷ gliotoxin,⁸ sporidesmin,⁹ chaetocin,¹⁰ chetomin,¹¹ and epicorazins¹² have comparable properties.

In addition to compounds 1 and 2, monocerin (4) was identified by spectroscopic methods as a major component of the EtOAc extract. Monocerin (4) is also a major phytotoxic compound of *Helminthosporium monoceras*.¹³

Experimental Section

^1H and ^{13}C NMR spectra were recorded with a Bruker WM-250 FT spectrometer. Chemical shifts were taken in δ units relative to Me_4Si (=0) with CDCl_3 as the solvent. Mass spectra

(5) Baute, R.; Deffieux, G.; Baute, M. A.; Filleau, M. J.; Neveu, A. *Tetrahedron Lett.* 1976, 3943–3944.

(6) Sugawara, F.; Strobel, G. A. *Plant Sci.*, in press.

(7) Nagarajan, R.; Huckstep, L. L.; Lively, D. H.; Delong, D. C.; Marsh, M. M.; Neuss, N. *J. Am. Chem. Soc.*, 1968, 90, 2980.

(8) Bell, M. R.; Johnson, J. R.; Wildi, B. S.; Woodward, R. B. *J. Am. Chem. Soc.* 1958, 80, 1001.

(9) Fridrichsons, J.; Mathieson, A. M. *Tetrahedron Lett.* 1962, 1265–1268.

(10) Hauser, D.; Weber, H. P.; Sigg, H. P. *Helv. Chim. Acta* 1970, 53, 1061–1073.

(11) Safe, S.; Tayler, A., *J. Chem. Soc., Perkin Trans. 1* 1972, 472–479.

(12) Baute, M. A.; Deffieux, G.; Baute, R.; Neveu, A. *Bull. Soc. Pharm. Bordeaux* 1980, 119, 203–210.

(13) Scott, F. E.; Simpson, T. J.; Trimble, L. A.; Vederas, J. E. *J. Chem. Soc., Chem. Commun.* 1984, 756–758 and references therein.

were obtained with a VG Analytical Model VG7070HE mass spectrometer. IR spectra were recorded on a Beckman IR-20 spectrophotometer and optical rotation was observed on a Perkin-Elmer Model 241MC polarimeter.

Toxin Production and Purification. The mycelium of *E. holmii*¹⁴ was inoculated into M-1-D culture broth¹⁵ and then shaken at 200 rpm for 2–3 weeks at 26 °C under luminescence (25 μ Einsteins M⁻² s⁻¹). Both the culture filtrate and the fluid obtained after centrifugation of the homogenized mycelium were extracted with EtOAc, and the extract was evaporated to dryness under reduced pressure and subjected to flash chromatography¹⁶ using (a) PhMe/EtOAc (2:1) and (b) CHCl₃/MeOH (70:1). On TLC (silica gel) the *R_f* values in CHCl₃/MeOH (50:1) were 0.19 for 1, 0.15 for 2, and 0.52 for monocerin (4). Both 1 and 2 could be isolated in mg quantities by reverse phase HPLC (Merck, RP-18) with CH₃CN/H₂O (63:35) as the solvent (1 mL/min). The retention time of both 1 and 2 in this HPLC system was 3.5 min, and that of 4 was 5.8 min. *E. holmii* produced a total of ca. 3.3 mg of 1, 0.2 mg of 2, and 75 mg of 4 in each liter of culture broth.

Exserohilone (1): colorless glass; [α]_D²¹ -247° (c 0.32, CHCl₃); IR (KBr) ν_{\max} 3320, 2910, 1680, 1640, 1395, 1260 cm⁻¹; EILRMS, *m/z* (relative intensity) 450 (2), 405 (24), 403 (100), 375 (27), 356 (27), 355 (28), 329 (9), 328 (18), 160 (6), 133 (8), 132 (8), 110 (5), 95 (7). EIHRMS, C₂₀H₂₂N₂O₆S₂ (M⁺; obsd *m/z* 450.0919, calcd *m/z* 450.0920) and C₁₉H₁₉N₂O₆S₁ (M⁺ - SMe; obsd *m/z* 403.0963, calcd *m/z* 403.0964); ¹H NMR (CDCl₃, 250 MHz) δ 2.26 (6 H, s, SMe), 2.23 (2 H, overlapping, d, *J* = 12.6, 13.6 Hz, H_a-7, H_a-14), 2.67 (2 H, d, d, *J* = 4.7, 13.6 Hz, H_e-7, H_e-14), 3.49 (2 H, d, d, *J* = 4.7, 12.6 Hz, H-7a, H-14a), 3.89 (2 H, d, d, *J* = 8.7, 12.6 Hz, H-4a, H-11a), 4.70 (2 H, d, d, *J* = 1.9, 1.9, 8.7 Hz, H-4, H-11), 6.10 (2 H, s, OH), 6.11 (2 H, d, d, *J* = 1.9, 9.1 Hz, H-3, H-10), 6.93 (2 H, d, d, *J* = 1.9, 9.1 Hz, H-2, H-9); ¹³C NMR (CDCl₃, 62.9 MHz) δ 15.1 (q, SMe), 32.1 (t, C-7, C-14), 46.3 (d, C-7a, C-14a), 69.0 (d, C-4a, C-11a), 72.4 (s, C-6a, C-13a), 72.9 (d, C-4, C-11), 128.6 (d, C-2, C-9) 151.1 (d C-3, C-10), 168.0 (s, C-6, C-13), 195.2 (s, C-1, C-8).

Acetate of Exserohilone. Exserohilone 1 (1.6 mg) was acetylated in the usual manner using dry pyridine and acetic anhydride to give the acetate in a quantitative yield: EILRMS, *m/z* (relative intensity) 534 (1.9, M⁺), 488 (21), 487 (30, M⁺ - SMe), 459 (6), 440 (22, M⁺ - 2 SMe), 439 (32), 397 (10), 379 (16), 367 (8), 160 (11), 133 (11), 132 (13), 94 (11); ¹H NMR (CDCl₃, 250 MHz) δ 2.129 (6 H, s, OAc), 2.132 (6 H, s, SMe), 2.11 (2 H, overlapping, d, *J* = 7.6, 13.6 Hz, H_a-7, H_a-14), 2.55 (2 H, d, d, *J* = 4.9, 13.6 Hz, H_e-7, H_e-14), 3.58 (2 H, d, d, *J* = 4.9, 7.6 Hz, 13.6, H-7a, H-14a), 4.08 (2 H, d, d, *J* = 8.6, 13.6 Hz, H-4a, H-11a), 5.89 (2 H, d, d, *J* = 1.8, 2.3, 8.6 Hz, H-4, H-11), 6.16 (2 H, d,

d, *J* = 2.3, 10.0 Hz, H-3, H-10), 6.70 (2 H, d, d, *J* = 1.8, 10.0 Hz, H-2, H-9).

4,11-Bis(*p*-bromobenzoate) of Exserohilone (3). Exserohilone 1 (2.0 mg) was *p*-bromobenzoyleated under standard conditions to give the bis(*p*-bromobenzoate) in a quantitative yield: ¹H NMR (CDCl₃, 250 MHz) δ 1.93 (6 H, s, SMe), 2.06 (2 H, d, d, *J* = 13.0, 13.7 Hz, H_a-7, H_a-14), 2.47 (2 H, d, d, *J* = 4.9, 13.7 Hz, H_e-7, H_e-14), 3.63 (2 H, d, d, *J* = 4.9, 13.0, 13.0 Hz, H-7a, H-14a), 4.27 (2 H, d, d, *J* = 8.8, 13.0 Hz, H-4a, H-11a), 6.08 (2 H, d, d, *J* = 1.9, 2.0, 8.8, H-4, H-11), 6.21 (2 H, d, d, *J* = 2.0, 10.3 Hz, H-3, H-10), 6.83 (2 H, d, d, *J* = 1.9, 10.3 Hz, H-2, H-9), 7.61 (4 H, d, *J* = 8.5 Hz, Ar), 7.98 (4 H, d, *J* = 8.5 Hz, Ar).

9,10-Dihydroexserohilone (2): [α]_D²¹ -110° (c 0.33 CHCl₃); EILRMS, *m/z* (relative intensity) 452 (1.8, M⁺), 406 (8), 405 (33, M⁺ - SMe), 359 (9), 358 (33, M⁺ - 2 SMe), 357 (100), 341 (5) 339 (7), 323 (6), 273 (13); EIHRMS, C₂₀H₂₄N₂O₆S₂ (M⁺, obsd *m/z* 452.1075, calcd *m/z* 452.1077) and C₁₉H₂₁N₂O₆S₁ (M⁺ - SMe, obsd *m/z* 405.1120, calcd *m/z* 405.1121); ¹H NMR (CDCl₃, 250 MHz) δ 1.6 (2 H, m, H-10), 1.96 (1 H, m, H-7), 2.238 (3 H, s, SMe), 2.242 (3 H, s, SMe), 2.27 (1 H, d, d, *J* = 12.0, 13.7 Hz, H_a-14), 2.51 (3 H, overlapping, m, H-7', H-9), 2.64 (1 H, d, d, *J* = 4.7, 13.7 Hz, H_e-14), 3.49 (2 H, m, H-7a, H-14a), 3.77 (1 H, d, d, *J* = 8.6, 13.9, H-11a), 3.86 (1 H, d, d, *J* = 8.6, 13.9 Hz, H-4a), 4.27 (1 H, d, d, *J* = ca. 6, 8.6 Hz, H-11), 4.69 (1 H, d, d, *J* = ca. 1.6, 2.0, 8.6 Hz, H-4), 5.66 (1 H, s, OH), 6.09 (1 H, d, d, *J* = 2.0, 10.1 Hz, H-3), 6.12 (1 H, s, OH), 6.92 (1 H, d, d, *J* = 1.6, 10.1 Hz, H-2).

Crystallographic Data and X-ray Structure Analysis of Bis(*p*-bromobenzoate) 3. All crystallographic calculations were done on a PRIME 850 computer operated by the Cornell Chemistry Computing Facility. Principal programs employed were as follows: REDUCE and UNIQUE, data reduction programs by M. E. Leonowicz, Cornell University, 1978; MULTAN 78, MULTAN 80, and RANTAN 80, systems of computer programs for the automatic solution of crystal structures from X-ray diffraction data (locally modified to perform all Fourier calculations including Patterson syntheses) written by P. Main, S. E. Hull, L. Lessinger, G. Germain, J. P. Declercq, and M. M. Woolfson, University of York, England, 1978 and 1980; BLS78A, an anisotropic block diagonal least-squares refinement written by K. Hirotsu and E. Arnold, Cornell University, 1980; PLUTO78, a crystallographic illustration program by W. D. S. Motherwell, Cambridge Crystallographic Data Centre, 1978; BOND, a program to calculate molecular parameters and prepared tables written by K. Hirotsu, Cornell University, 1978.

Acknowledgment. K.S., F.S., and G.A.S. thank the Montana Agricultural Experiment Station for financial support. Y.F., H. C.-H., and J.C. thank NSF #INT-14133 and NIH #CA-24487 for financial assistance.

Supplementary Material Available: Tables of fractional coordinates, thermal parameters, interatomic distances, and interatomic angles for compound 3 (5 pages). Ordering information is given on any current masthead page.

(14) *E. holmii* was kindly provided by Dr. E. S. Luttrell, University of Georgia, Athens.

(15) Pinkerton, F.; Strobel, G. A. *Proc. Natl. Acad. Sci. U.S.A.* 1976, 73, 4007–4011.

(16) Still, W. C.; Kahn, M.; Mitra, A. *J. Org. Chem.* 1978, 43, 2923–2925.